

DATE:

NAME:

CLASS:

CHAPTER 1	Binary Molecular Compound Nomenclature and Formulas Quiz	BLM 1.0.10
ASSESSMENT		

### Names and Formulas of Binary Molecular Compounds

$\text{ClF}_3$	
$\text{N}_2\text{F}_4$	
	dichlorine pentoxide
	dihydrogen dioxide (hydrogen peroxide)
$\text{N}_2\text{O}$	
$\text{NH}_3$	
	tetraphosphorus decoxide
	sulfur trioxide
$\text{CCl}_4$	
$\text{SeF}_6$	
	silicon dioxide
	iodine pentafluoride
$\text{SI}_2$	
$\text{SiH}_4$	
	phosphorus pentasulfide
	boron trichloride
$\text{SO}_2$	
$\text{NO}_2$	
	dichlorine monoxide
	xenon tetroxide