

DATE:

NAME:

CLASS:

|            |                          |           |
|------------|--------------------------|-----------|
| CHAPTER 1  | Simple Binary Ionic Quiz | BLM 1.0.7 |
| ASSESSMENT |                          |           |
|            |                          |           |

### Formulas and Names of Simple Binary Ionic Compounds

|                         |                     |
|-------------------------|---------------------|
| $\text{SrCl}_2$         |                     |
| $\text{Li}_2\text{O}$   |                     |
|                         | potassium iodide    |
|                         | silver sulfide      |
| $\text{Be}_3\text{N}_2$ |                     |
| $\text{NaBr}$           |                     |
|                         | lithium hydride     |
|                         | radium selenide     |
| $\text{Al}_2\text{S}_3$ |                     |
| $\text{Zn}_3\text{P}_2$ |                     |
|                         | rubidium chloride   |
|                         | sodium nitride      |
| $\text{Ag}_3\text{P}$   |                     |
| $\text{BaF}_2$          |                     |
|                         | aluminium bromide   |
|                         | magnesium phosphide |
| $\text{CsI}$            |                     |
| $\text{AlP}$            |                     |
|                         | magnesium sulfide   |
|                         | cesium oxide        |