

**When drawing alkenes**

- The double bond appears after the carbon atom with the number stated in the suffix.
- The two carbon atoms sharing the double bond each have only two other bonds.

**Sample Problem**

Draw a structural formula for 2-methylbut-2-ene.

**Solution**

**Step 1** The root is *but-* so there are four carbon atoms in the main chain.

**Step 2** The suffix is *-2-ene* so it has a double bond after carbon atom number two.

**Step 3** The prefix is 2-methyl- so there is a  $\text{-CH}_3$  group of the second carbon atom.

**Step 4** All carbon atoms must be bonded to enough hydrogen atoms to give them exactly four bonds.

The structural formula of 2-methylbut-2-ene is:

