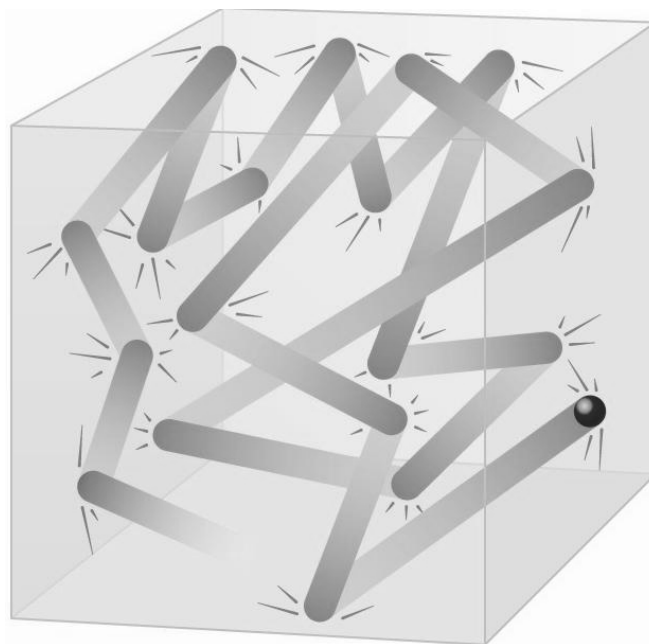


Ideal gas molecules move with random motion, colliding with one another and with the walls of the container.



The path of an individual molecule follows straight lines between collisions. Such a path is sometimes called “random walk.”