



A titration setup

Additional Titration Terminology

Equivalence point	stage of titration at which stoichiometrically equivalent amounts of reactant have been consumed
Endpoint	the point at which the indicator changes colour
Standardization	determination of the concentration of a solution by analyzing its reaction with a solution of known concentration (a standard)
pH	the negative logarithm (base-ten) of the molar concentration of hydronium ions in a solution
Indicator	a substance that changes colour over a given pH range; usually a weak, monoprotic acid
Titration Curve	a plot of the pH of a reaction mixture versus the volume of titrant added